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5. 35 g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
Determine the number
of grams in each of the
quantities below. 1, 2.5
moles of NaCl 2. 0.50
moles of 3, 1.70 moles
of KMnO_4 4, 0.25 moles
of KCl e, ST 5. 3.2
moles of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
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Name How many
grams of potassium
chloride are produced
if 25 g of potassium
chlorate decompose? /
25 ken How many
grams of hydrogen are
necessary to react
completely with 50.0 g
of nitrogen in the
above reaction? 3, How
many grams of

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ammonia are produced
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r i c Page 4:2 ANSWER

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NAMING

HYDROCARBONS Name

Name the compounds
below according to the
IUPAC naming system

1. $\begin{array}{ccccccc} \text{H} & \text{H} & \text{H} & | & | & | & \text{H} \\ | & | & | & | & | & | & | \\ \text{---C} & \text{---C} & & & & & \\ | & & & & & & \\ \text{---C} & \text{---H} & | & | & | & \text{H} & \text{H} & \text{H} \end{array}$ 5.

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